



QP CODE: 23004203

Reg No :

M Sc DEGREE (CSS) EXAMINATION, JUNE 2023

Fourth Semester

M Sc CHEMISTRY

Elective - CH800403 - ADVANCED PHYSICAL CHEMISTRY

2019 ADMISSION ONWARDS

2CB5253D

Time: 3 Hours

Weightage: 30

Part A (Short Answer Questions)

Answer any **eight** questions.

Weight **1** each.

- 1. Explain the important features of thermoluminescence.
- 2. Give the important advantages of cadmium telluride solar cells.
- 3. Define a fluorophore. Give three examples.
- 4. Explain the principle of electron diffraction analysis.
- 5. Explain ion association.
- 6. Explain Liquid junction potential.
- 7. Describe over potential.
- 8. Explain about a typical cyclic voltammogram.
- 9. What are the advantages of polarography?
- 10. What is thermal osmosis? Explain briefly.

(8×1=8 weightage)

Part B (Short Essay/Problems)

Answer any **six** questions.

Weight **2** each.

- 11. Derive the Stern Volmer equation.
- 12. Write short notes on the following parts of a fluorescent spectrometer (a) monochromators (b) polarizers.
- 13. Discuss in detail about the structure and working of HCL.



- Discuss the salient features of Helmoltz and Stern double layers.
- 15. Write a note on methyl alcohol fuel cells.
- 16. Explain about decomposition potential and supporting electrolyte used in polarography.
- 17. Discuss about the different coulometric analysis with constant current?
- 18. Radiation of wavelength 2200 A⁰ is passed through a cell containing 20 mL of solution which contains 0.05 M oxalic acid and 0.001M uranylnitrate. After absorption of 90J of radiation energy, the concentration of oxalic acid is reduced to 0.03 M. Calculate the quantum yield for the decomposition.

(6×2=12 weightage)

Part C (Essay Type Questions)

Answer any **two** questions.

Weight **5** each.

- 19. Write short notes on (a) application of lasers in photochemical kinetic study. (b) two photon absorption spectroscopy.
- 20. Give an account of the principle and instrumentation of FES. Discuss the applications of FES briefly.
- 21. Derive DHO equation. What are the main drawbacks of DHO?
- 22. What is the theory of amperometric titrations? Discuss about the application of amperometry in the qualitative analysis of anions and cations in solution?

(2×5=10 weightage)