

OP CODE: 19102431



Reg No

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Name

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BSc DEGREE (CBCS) EXAMINATION, OCTOBER 2019

Fifth Semester

Core Course - CH5CRT07 - PHYSICAL CHEMISTRY - I

B.Sc Chemistry Model I ,B.Sc Chemistry Model II Industrial Chemistry ,B.Sc Chemistry Model III

Petrochemicals

2017 Admission Onwards

99903D5F

Maximum Marks: 60

Time: 3 Hours

Part A

Answer any ten questions.

Each question carries 1 mark.

- 1. What is meant by compressibility factor?
- 2. For a given sample of gas what is the order of RMS velocity, average velocity and most probable velocity?
- 3. Define mean free path.
- 4. Write down Chapman equation.
- 5. Why H2O is liquid whereas H2S exists as gas at room temperature?
- 6. How many atoms per unit cell are present in fcc lattice?
- 7. Write two examples for ionic crystals having AX symmetry.
- 8. What is the main application of impurity defect?
- 9. Write two examples for cholesteric liquid crystals.
- 10. What is chemisorption?
- 11. What is the importance of BET equation?
- 12. What is meant by critical micelle concentration?

 $(10 \times 1 = 10)$

Part B

Answer any six questions.

Each question carries 5 marks.

13. Give the postulates of kinetic theory of gases.



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- 14. Define critical temperature, critical pressure and critical volume. How are they related to Van der Waal's constant a and b?
- 15. Obtain the virial form of Van der Waal's equation.
- 16. What is surface tension? How is it determined?
- 17. Briefly explain the Debye Scherrer method for the analysis of crystal structures.
- 18. Compare the structure of NaCl and KCl by using Powder method.
- 19. Account for the yellow colour of NaCl when heated in sodium vapour.
- 20. How is the surface area of adsorbent determined from BET theory?
- 21. Write a note on Brownian movement and Tyndall effect.

 $(6 \times 5 = 30)$

Part C

Answer any two questions.

Each question carries 10 marks.

- 22. Derive Van der Waal's equation of real gases. How is it expressed in virial form?
- 23. What is meant by coefficient of viscosity? How is viscosity determined using Ostwald viscometer?
- 24. (a) Explain the Laws of Crystallography. (b) Write a short note on Weiss indices and Miller indices.
- 25. a) Discuss any two methods on how colloidal solutions are purified. b) Explain the terms micelles and critical micelle concentration.

 $(2 \times 10 = 20)$

