

QP CODE: 19002497



Reg No :	
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Name :

M.Sc. DEGREE (C.S.S) EXAMINATION, NOVEMBER 2019 First Semester

Faculty of Science

Core - CH500103 - QUANTUM CHEMISTRY AND GROUP THEORY

(Common to all Branches of Chemistry)
2019 Admission Onwards
9F63094D

Time: 3 Hours Maximum Weight :30

Part A (Short Answer Questions)

Answer any **eight** questions.

Weight **1** each.

- Prove that $S_2 = i$ for molecular point groups.
- Write a note on Hermann Mauguin symbols.
- 3. " A square of any element is also an element in the same group". Illustrate.
- 4. Show that inversion operation and rotation operation commute each other.
- 5. What are block factored matrices?
- 6. What is meant by black body radiation? Explain.
- 7. Determine the average value of linear momentum for particle in a one dimensional box.
- 8. Explain the term degeneracy. What is the maximum degeneracy possible for a particle in a cube ?
- 9. Write the equations for converting Cartesian co-ordinates to the spherical polar co-ordinates.
- 10. What are slater determinants? Write the slater determinant for Berilium atom.

(8×1=8 weightage)

Part B (Short Essay/Problems)

Answer any six questions.

Weight 2 each.

11. Distinguish between Dnh and Dnd point groups



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- 12. Construct GMT for C2v and C2h point groups.
- 13. How does the concept of irreducible representations help in the structure ellucidation of molecules?
- 14. Construct SALCs of BF3 molecule.
- 15. Show that the eigenvalues of a Hermitian operator are real.
- 16. Evaluate $[L_yL_x]$.
- 17. What are spherical Harmonics? Determine the first three spherical harmonics.
- 18. What are spin orbitals? How will you construct spin orbitals?

(6×2=12 weightage)

Part C (Essay Type Questions)

Answer any two questions.

Weight 5 each.

- 19. What do you mean by similarity transformation and classes in group theory? Illustrate using C3v point group.
- 20. Construct the character table for C3v point group.
- Set up Schrodinger equation and find eigen values and eigen functions for a particle moving on a ring.
- Write down the Schrodinger equation for the electron in a H atom. Transform the equation into spherical polar co-ordinates and separate into q equation, Fequation and R equation. what are the solutions for each?

(2×5=10 weightage)

